

Periodensystem der Elemente

Periodic Table of the Elements

1 H Wasserstoff Hydrogen [He] 1s [1,0078; 1,0082] -1,1 13,99* / 20,271* 0,084 2,2 / 7,18	2 He Helium Helium 1s ² 4,0026 0 - / 4,222 0,170	3 Li Lithium Lithium [He] 2s [6,938; 6,997] 1 453,65 / 1603 530 0,98 / 3,01	4 Be Beryllium Beryllium [He] 2s ² [9,0122 2 1560 / 2742 1850 1,57 / 4,9	5 B Bor Bor [He] 2s ² 2p [10,806; 10,821] 3 2349 / 4200 2460 2,04 / 4,29	6 C Kohlenstoff Carbon [He] 2s ² 2p [12,0006; 12,012] -4,-3,-2,-1,0,1,2,3,4 - / 3915* 54,3* / 90,188* 1,33 3,44 / 7,54	7 N Stickstoff Nitrogen [He] 2s ² 2p ³ [14,0064; 14,008] -3,-2,-1,1,2,3,4,5 63,15* / 77,355* 1,170 3,04 / 7,3	8 O Sauerstoff Oxygen [He] 2s ² 2p ⁴ [15,999; 16,000] -2,-1 54,3* / 90,188* 1,33 3,44 / 7,54	9 F Fluor Fluorine [He] 2s ² 2p ⁵ [18,998 20,180 0 26,56 / 27,104 0,84	10 Ne Neon Neon [He] 2s ² 2p ⁶ [20,180 20,180 0 26,56 / 27,104 0,84	11 Na Natrium Sodium [Ne] 3s [22,990 1 370,944 / 1156,09 970 0,93 / 2,85	12 Mg Magnesium Magnesium [Ne] 3s ² [24,304; 24,307] 2 923 / 1363 1740 1,31 / 3,75	13 Al Aluminium Aluminium [Ne] 3s ² 3p [26,982 3 933,47 / 2743 1820 1,61 / 3,23	14 Si Silicium Silicon [Ne] 3s ² 3p ² [28,084; 28,086] -4,4 1687 / 3538 2320 1,9 / 4,77	15 P Phosphor Phosphorus [Ne] 3s ² 3p ³ [30,974 3 317,3* / 550** 2320 2,19 / 5,62	16 S Schwefel Sulfur [Ne] 3s ² 3p ⁴ [32,059; 32,076] -2,2,4,6 368,4* / 717,8* 2320 2,58 / 6,22	17 Cl Chlor Chlorine [Ne] 3s ² 3p ⁵ [35,446; 35,457] -1,1,3,5,7 171,6* / 239,11* 2320 3,16 / 8,3	18 Ar Argon Argon [Ne] 3s ² 3p ⁶ [39,948 39,948 0 83,81 / 87,302 1,66	19 K Kalium Potassium [Ar] 4s [39,098 1 336,7 / 1032 860 0,82 / 2,42	20 Ca Calcium Calcium [Ar] 3d ⁴ 4s [40,078(4) 2 1115 / 1757 1540 1 / 2,2	21 Sc Scandium Scandium [Ar] 3d ⁴ 4s [44,956 3 1814 / 3109 2990 1,36 / 3,34	22 Ti Titan Titanium [Ar] 3d ² 4s [47,867 4 2,3,4 1941* / 3560* 4510 1,54 / 3,45	23 V Vanadium Vanadium [Ar] 3d ³ 4s [50,942 2 0,2,3,4,5 2183 / 3680 6090 1,63 / 3,6	24 Cr Chrom Chromium [Ar] 3d ⁵ 4s [51,996 4 0,2,3,4,5,6 2180 / 2755 7440 1,66 / 3,72	25 Mn Mangan Manganese [Ar] 3d ⁵ 4s [54,938 5 -1,0,2,3,4,6 1519 / 2334 7440 1,55 / 3,72	26 Fe Eisen Iron [Ar] 3d ⁶ 4s [55,845(2) 2 0,2,3,4 1768 / 3200 8890 1,83 / 4,06	27 Co Cobalt Cobalt [Ar] 3d ⁷ 4s [58,933 2 -1,0,2,3 1728 / 3003 8910 1,88 / 4,3	28 Ni Nickel Nickel [Ar] 3d ⁸ 4s [58,693 2 0,2,3,4 1728 / 3003 8910 1,91 / 4,4	29 Cu Kupfer Copper [Ar] 3d ¹⁰ 4s [63,546(3) 2 1,2 1357,77 / 2835 8920 1,9 / 4,48	30 Zn Zink Zinc [Ar] 3d ¹⁰ 4s [65,38(2) 2 65 692,68 / 1180 7140 1,65 / 4,45	31 Ga Gallium Gallium [Ar] 3d ¹⁰ 4s [69,723 3 302,9146 / 2673 5910 1,81 / 3,2	32 Ge Germanium Germanium [Ar] 3d ¹⁰ 4s [72,630(8) 2 -3,5 1211,4 / 3106 5320 2,01 / 4,6	33 As Arsen Arsenic [Ar] 3d ¹⁰ 4s [74,922 3 -3,5 1090 / 887* 5910 2,18 / 5,3	34 Se Selen Selenium [Ar] 3d ¹⁰ 4s [78,971(8) 2 -2,4,6 494* / 958* 5720 2,55 / 5,89	35 Br Brom Bromine [Ar] 3d ¹⁰ 4s [79,901; 79,907] 3 -1,1,3,5,7 265,8* / 332* 3140 2,96 / 7,59	36 Kr Krypton Krypton [Ar] 3d ¹⁰ 4s [83,798(2) 0,2 115,78 / 119,735 3,48	37 Rb Rubidium Rubidium [Kr] 5s [85,468 1 312,45 / 961 1530 0,82 / 2,34	38 Sr Strontium Strontium [Kr] 5s [87,62 2 1050 / 1650 2630 0,95 / 2	39 Y Yttrium Yttrium [Kr] 4d ⁵ 5s [88,906 3 1799 / 3203 4470 1,22 / 3,19	40 Zr Zirkonium Zirconium [Kr] 4d ⁵ 5s [91,224(2) 2 2,3,4 2128 / 4650 6510 1,33 / 3,64	41 Nb Niob Niobium [Kr] 4d ⁵ 5s [92,906 3,5 0,2,3,4,5,6 2750 / 5017 8580 1,6 / 4,0	42 Mo Molybdän Molybdenum [Kr] 4d ⁵ 5s [95,95 4 0,2,3,4,5,6 2896 / 4912 11490 2,16 / 3,9	43 Tc Technetium Technetium [Kr] 4d ⁵ 5s [97] 4,7 0,2,4 2430 / 4538 11490 1,9	44 Ru Ruthenium Ruthenium [Kr] 4d ⁷ 5s [101,07(2) 2 0,2,4 2607 / 4423 12450 2,28 / 4,3	45 Rh Rhodium Rhodium [Kr] 4d ⁸ 5s [102,91 2 0,2,4 2237 / 3968 12450 2,28 / 4,3	46 Pd Palladium Palladium [Kr] 4d ¹⁰ 106,42 107,87 124,93 / 2483 12020 2,2 / 4,45	47 Ag Silber Silver [Kr] 4d ¹⁰ 5s [107,87 1 107,87 1234,93 / 2483 12020 1,93 / 4,44	48 Cd Cadmium Cadmium [Kr] 4d ¹⁰ 5s [112,41 2 112,41 594,22 / 1040 8640 1,69 / 4,33	49 In Indium Indium [Kr] 4d ¹⁰ 5s [114,82 3 1,3 429,75 / 2345 7310 1,78 / 3,1	50 Sn Zinn Tin [Kr] 4d ¹⁰ 5s [118,71 2 -2,4,6 505,08* / 2875* 6690 1,96 / 4,3	51 Sb Antimon Antimony [Kr] 4d ¹⁰ 5s [121,76 3 -3,5 903,78 / 1908 12450 2,05 / 4,85	52 Te Tellur Tellurium [Kr] 4d ¹⁰ 5s [127,60(3) 2 -2,4,6 722,66 / 1261 6940 2,02 / 4,69	53 I Jod Iodine [Kr] 4d ¹⁰ 5s [126,90 2 -1,1,3,5,7 722,66 / 1261 6940 2,66 / 6,76	54 Xe Xenon Xenon [Kr] 4d ¹⁰ 5s [131,29 0,2 115,78 / 119,735 4,49	55 Cs Cäsium Caesium [Xe] 6s [132,91 1 301,7 / 944 1900 0,79 / 2,18	56 Ba Barium Barium [Xe] 6s [137,33 2 1000 / 1910 3650 0,89 / 2,4	57 La Lanthan Lanthanum [Xe] 5d ⁶ s ² [138,91 3 1193 / 3737 16680 1,1 / 3,1	58 Ce Cer Cerium [Xe] 4f ¹ 5d ¹ 6s ² [140,12 3,4 1068 / 3716 6770 1,12 / -	59 Pr Praseodym Praseodymium [Xe] 4f ³ 6s ² [140,91 3,4 1208 / 3403 6480 - / 1,13	60 Nd Neodym Neodymium [Xe] 4f ⁴ 6s ² [144,24 3,4 1297 / 3347 7000 1,14 / -	61 Pm Promethium Promethium [Xe] 4f ⁵ 6s ² [145] 3 1315 / 3273 7220	62 Sm Samarium Samarium [Xe] 4f ⁶ 6s ² [150,36(2) 3,4 1345 / 2173 7540 1,17 / -	63 Eu Europium Europium [Xe] 4f ⁷ 6s ² [151,96 3,4 1099 / 1802 7890 1,2 / -	64 Gd Gadolinium Gadolinium [Xe] 4f ⁷ 5d ¹ 6s ² [157,25(3) 3,4 1629 / 3396 7890 1,2 / -	65 Tb Terbium Terbium [Xe] 4f ⁹ 6s ² [158,93 3,4 1629 / 3396 8250	66 Dy Dysprosium Dysprosium [Xe] 4f ¹⁰ 6s ² [162,50 3,4 1680 / 2840 8560 1,22 / -	67 Ho Holmium Holmium [Xe] 4f ¹¹ 6s ² [164,93 3,4 1734 / 2873 8780 1,24 / -	68 Er Erbium Erbium [Xe] 4f ¹² 6s ² [167,26 3,4 1818 / 3141 9050 1,24 / -	69 Tm Thulium Thulium [Xe] 4f ¹³ 6s ² [168,93 3,4 1818 / 2223 9320 1,27 / -	70 Yb Ytterbium Ytterbium [Xe] 4f ¹⁴ 6s ² [173,05 3 1925 / 3675 9840 1,27 / -	71 Lu Lutetium Lutetium [Xe] 4f ¹⁴ 5d ¹ 6s ² [174,97 3 1925 / 3675 9840 1,27 / -	Elektronenkonfiguration relative Atommasse wichtige Oxidationszahlen Schmelz- / Siedepunkt in Kelvin Dichte in kg/m ³ (bei 293 K) Elektroneg. Pauling/Pearson	72 Yt Ytterbium Ytterbium [Xe] 4f ¹⁴ 6s ² [173,05 3 1925 / 3675 9840 1,27 / -	73 Ts Tennessine Tennessine [Xe] 4f ¹⁴ 6s ² 7p ⁵ [293] 1 1	74 Og Oganesson Oganesson [Xe] 4f ¹⁴ 6s ² 7p ⁶ [294] 0	75 Fr Francium Francium [Rn] 7s [223] 1 300 / (ca.) 950 348	76 Ra Radium Radium [Rn] 7s ² [226] 2 973 / 2010 5500	77 Ac Actinium Actinium [Rn] 6d ⁷ s ² [227] 3 (ca.) 1323 / 3471 10070	78 Rf Rutherfordium Rutherfordium [Rn] 5f ¹⁴ 6d ⁷ 7s ² [267] 4 (vermutet)	79 Db Dubnium Dubnium [Rn] 5f ¹⁴ 6d ⁷ 7s ² [270] 3,4,5 (vermutet)	80 Sg Seaborgium Seaborgium [Rn] 5f ¹⁴ 6d ⁷ 7s ² [271] 6 (vermutet)	81 Bh Bohrium Bohrium [Rn] 5f ¹⁴ 6d ⁷ 7s ² [270] 7	82 Hs Hassium Hassium [Rn] 5f ¹⁴ 6d ⁷ 7s ² [278] 8 (vermutet)	83 Mt Meitnerium Meitnerium [Rn] 5f ¹⁴ 6d ⁷ 7s ² [278] 3,4,6 (vermutet)	84 Ds Darmstadtium Darmstadtium [Rn] 5f ¹⁴ 6d ⁷ 7s ² [281] 2,4,6 (vermutet)	85 Rg Roentgenium Roentgenium [Rn] 5f ¹⁴ 6d ⁷ 7s ² [281] 3 (vermutet)	86 Cn Copernicium Copernicium [Rn] 5f ¹⁴ 6d ⁷ 7s ² [285] 2	87 Nh Nihonium Nihonium [Rn] 5f ¹⁴ 6d ⁷ 7s ² 7p ¹ [286] 3	88 Fl Flerovium Flerovium [Rn] 5f ¹⁴ 6d ⁷ 7s ² 7p ² [289] 4	89 Mc Moscovium Moscovium [Rn] 5f ¹⁴ 6d ⁷ 7s ² 7p ³ [289] 3	90 Lv Livermorium Livermorium [Rn] 5f ¹⁴ 6d ⁷ 7s ² 7p ⁴ [293] 2	91 Ts Tennessine Tennessine [Rn] 5f ¹⁴ 6d ⁷ 7s ² 7p ⁵ [293] 1	92 Og Oganesson Oganesson [Rn] 5f ¹⁴ 6d ⁷ 7s ² 7p ⁶ [294] 0	93 Np Neptunium Neptunium [Rn] 5f ⁷ 6d ¹ 7s ² [237] 3,4,5,6 917 / 4273 20480 1,36 / -	94 Pu Plutonium Plutonium [Rn] 5f ⁶ 6d ² 7s ² [244] 3,4,5,6 912,5 / 3501 19740 1,28 / -	95 Am Americium Americium [Rn] 5f ⁷ 7s ² [243] 3,4,5,6 1449 / 2880 13670 1,3 / -	96 Cm Curium Curium [Rn] 5f ⁸ 6d ² 7s ² [247] 3,4 1613 / 3383 13510 1,3 / -	97 Bk Berkelium Berkelium [Rn] 5f ⁹ 7s ² [247] 3,4 1323* / - 13250 1,3 / -	98 Cf Californium Californium [Rn] 5f ¹⁰ 7s ² [251] 3 1173 / - 15100 1,3 / -	99 Es Einsteinium Einsteinium [Rn] 5f ¹¹ 7s ² [252] 3 1133 / - 15100 1,3 / -	100 Fm Fermium Fermium [Rn] 5f ¹² 7s ² [257] 3 1800 / - 15100 1,3 / -	101 Md Mendelevium Mendelevium [Rn] 5f ¹³ 7s ² [258] 3 1100 / - 15100 1,3 / -	102 No Nobelium Nobelium [Rn] 5f ¹⁴ 7s ² [259] 3 1100 / - 15100 1,3 / -	103 Lr Lawrencium Lawrencium [Rn] 5f ¹⁴ 7s ² 7p ¹ [260] 3 1900 / - 15100 1,3 / -	electron configuration atomic weight important oxidation numbers melting / boiling point in Kelvin density in kg/m ³ (at 293 K) electroneg. Pauling / Pearson
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